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CHAPTER 11 REVIEW Gases Class SHORT

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ANSWER Answer the following questions in the space provided.

c c The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (b) divided by the mass of 1 mol. nRT (c) multiplied by 22.4 L. (d) divided by 22.4 L. For the expression $V =$ (a) increasing P (b) decreasing T

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Describe two ways that atmospheric

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nitrogen is changed to form useful compounds. nitrogen—fixing bacteria, Haber process to make NH_3 Chapter 11 Typical Elements 137 Chapter Review 25. 26. 138 In 1669, Hemiig Brandt noticed a new substance was produced when he heated urine and sand together in an attempt to make gold.

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TYPICAL, \ufb02 ELEMENTS 11.1 ...

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Intermolecular force. Dipole-dipole force.
Hydrogen bond. Dispersion force
(London force) an electrostatic attraction
between molecules; it is much weak....

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the attraction of the positive end of one polar molecule to th.... a special dipole-dipole force involving hydrogen and a highly....

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Answer: A 11) A substance that can't be

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chemically broken down into simpler substances is A) a homogeneous mixture. B) an element. C) a heterogeneous mixture. D) a compound. E) an electron. Answer: B

12) A substance composed of two or more elements in a fixed, definite proportion is A) a homogeneous mixture. B) a heterogeneous mixture. C) a compound.

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General Chemistry 101 Chapter 1 Review Questions

Chapter 1 Review Solutions: 11 The mass has four significant figures, while the density has three significant figures, so the answer must have three significant figures. 22. It is silver. To determine whether or not it is silver, we

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can calculate its density. However, the

Answer Key and Tests for Discovering Design with Sample ...

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